

Cool and hot science for a bright future

Science in School is published by EIROforum, a collaboration between eight of Europe's largest inter-governmental scientific research organisations. This article reviews some of the latest news from the EIROforum members (EIROs).



CERN: proudly presenting the new (Higgs?) boson

On 4 July 2012, the ATLAS and CMS experiments at CERN's Large Hadron Collider (LHC) announced the discovery of a new particle with a mass of about 126 GeV and characteristics similar to those expected for the long-sought Higgs boson. This discovery is a major step forward in our understanding of the origin of mass, and of the fundamental properties of matter, space and time.

The preliminary results come from data taken in 2011 and 2012, corresponding to about 10^{15} proton-proton collisions at 7 and 8 TeV energy. More data will be needed to study the different decay modes of the new particle, to find out if the theoretical predictions are correct, or if there are any – even tiny – differences between the predictions of the Higgs model and the observations.

Find out more about the search for the Higgs in this issue's feature article:

Hayes E (2012) Accelerating the pace of science: interview with CERN's Rolf Heuer. *Science in School* 25: 6-12. www.scienceinschool.org/2012/issue25/heuer

To learn more about how the LHC works and the search for the Higgs boson, see:

Landua R, Rau M (2008) The LHC: a step closer to the Big Bang. *Science in School* 10: 26-33. www.scienceinschool.org/2008/issue10/lhchw

Landua R (2008) The LHC: a look inside. *Science in School* 10: 34-45. www.scienceinschool.org/2008/issue10/lhchow

Based in Geneva, Switzerland, CERN is the world's largest particle physics laboratory. To learn more, see: www.cern.ch

For a list of CERN-related articles in *Science in School*, see: www.scienceinschool.org/cern

One of the proton-proton collisions at a centre of mass energy of 8 TeV, recorded with the CMS detector, which contributed to the discovery of the new particle.



EFDA-JET: paving the tungsten-tile road to ITER

ITER, the world's biggest fusion experiment, is under construction in the south of France. During its construction, the Joint European Torus (JET) – currently Europe's largest fusion experiment – is a vital test bed for ITER design and operation.

In this capacity, JET has successfully completed its 2011-2012 experimental campaign by running 151 identical high-powered plasma pulses, totalling 900 seconds of stable operating time. This emulates a single pulse of ITER – by virtue of its superconducting magnets, ITER will be able to maintain a stable pulse 20 times longer than JET. On the basis of the recent experiments, scientists are confident that they will be able to maintain stable fusion conditions for the length of these longer pulses. The experiment also tested the long-term behaviour of the wall materials – a range of JET's wall tiles will now be extracted by the remote-handling system (pictured) for analysis.

The ITER design team are very interested in the results, and have already requested an even more severe test for the materials when experiments recommence next year: deliberate melting of some tungsten wall tiles.

Situated in Culham, UK, JET is Europe's fusion device. Scientific exploitation of JET is undertaken through the European Fusion Development Agreement (EFDA). To learn more, see: www.efda.org

For a list of EFDA-JET-related articles in *Science in School*, see: www.scienceinschool.org/efdajet

The remote handling mascot is mounted on a 6 m articulated boom, which allows maintenance without humans entering the JET vessel. An operator in a remote control room controls the mascot's two gripper arms, which can operate around 1500 bespoke tools.

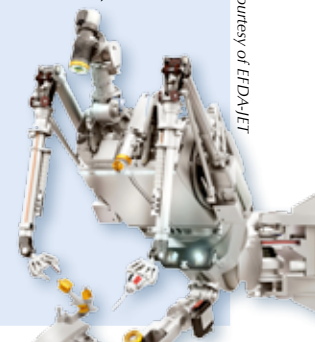


Image courtesy of EFDA/JET

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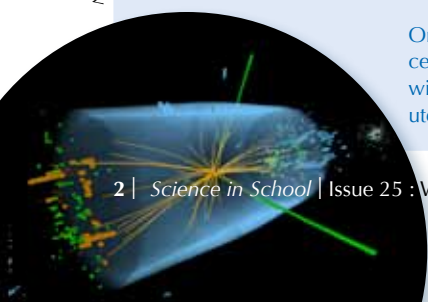


Image courtesy of EMBL-EBI

ENCODE researchers found that most of our DNA has a function: controlling when and where genes are turned on and off.

EMBL: ENCODEing the switchboard of the human genome

For the past five years, hundreds of scientists in the 'Encyclopaedia of DNA elements' (ENCODE) project have been systematically exploring what the human genome does, to identify all its functional elements. Their findings show that much of what has been called 'junk DNA' is actually a massive, 3D switchboard turning genes on and off. Together with colleagues from around the globe, scientists at EMBL's European Bioinformatics Institute have found that while only 2% of our DNA is genes, a much bigger part of the genome – at least 20% – is involved in controlling when and where those genes are active, and as much as 80% of the genome has a distinct biochemical activity. This opens up new avenues of biomedical research.

To give some sense of the scale of the project, ENCODE used around 300 years' worth of computer time studying 147 tissue types to determine what turns specific genes on and off, and how that 'switch' differs between cell types. All of the data is now publicly available and the findings are published in 30 connected, open-access papers in three science journals: *Nature*, *Genome Biology* and *Genome Research*.

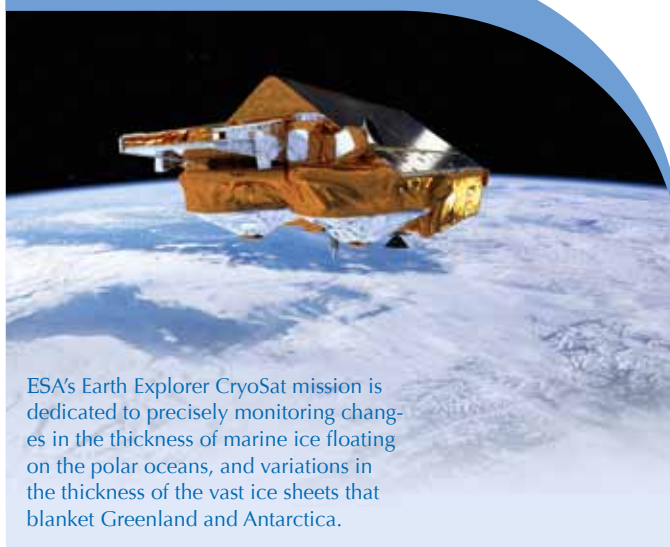
For more details including the full publication list, see the press release on the EMBL website (www.embl.org) or use the direct link: <http://tinyurl.com/ebiencode>

What is the ENCODE project all about? Watch a video interview with leading scientists Ewan Birney, Tim Hubbard and Roderic Guigo: <http://bit.ly/Ro9UDt>

Learn how the insights from the ENCODE project were represented in the 'Dance of the DNA'. Visit the website of the Science Museum, London, UK (www.sciencemuseum.org.uk) or use the direct link: <http://tinyurl.com/8cr7249>

EMBL is Europe's leading laboratory for basic research in molecular biology, with its headquarters in Heidelberg, Germany. The European Bioinformatics Institute is part of EMBL and is based in Cambridge, UK. To learn more, see: www.embl.org

For a list of EMBL-related articles in *Science in School*, see: www.scienceinschool.org/embl



ESA's Earth Explorer CryoSat mission is dedicated to precisely monitoring changes in the thickness of marine ice floating on the polar oceans, and variations in the thickness of the vast ice sheets that blanket Greenland and Antarctica.

ESA: The Arctic ice cap is thinning

The year 2012 saw the surface area of Arctic sea ice hit a record low since satellite measurements began in the 1970s. The consequence of losing part of the Arctic's ice coverage could be profound: the ice cap reflects sunlight back into space; sunlight that, unless reflected, would contribute to global warming.

The European Space Agency (ESA)'s satellites SMOS and CryoSat have found that not only is the area of sea-ice getting smaller but the ice is also getting thinner: 900 km³ of summer sea ice have disappeared from the Arctic ocean over the past year, a rate of loss that is 50% higher than most scenarios outlined by polar scientists.

In addition to the total ice volume, it is important to evaluate the thickness of the young ice that forms in winter. Only those areas thick enough to survive the next summer's melting period can become the basis of the following winter's thick ice. It is this thick, multi-year ice that ultimately indicates how healthy the Arctic is.

ESA's measurements show that this newly formed ice is becoming significantly thinner each year, so that less and less of it survives the summer. In particular, SMOS detected extensive areas less than half a metre thick. Scientists therefore predict that of the total Arctic sea-ice cover for Winter 2012-13, a larger fraction than ever before (about 12 million km²) will consist of thin ice. This suggests that less Arctic sea-ice than ever before will survive the melting phase in Summer 2013.

To model the effect of the changing ice cover with your students in class, see:

Shallcross D, Harrison T (2008) Climate change modelling in the classroom. *Science in School* 9: 28-33.
www.scienceinschool.org/2008/issue9/climate

See our article series on climate change in several languages: www.scienceinschool.org/climatechange

ESA is Europe's gateway to space, with its headquarters in Paris, France. For more information, see: www.esa.int

For a list of ESA-related articles in *Science in School*, see: www.scienceinschool.org/esa

Biology

Chemistry

Physics

Image courtesy of ESA / AOE's Medialab



ESO: Sweet discovery in space

Astronomers using ALMA, one of the world's largest ground-based astronomy projects, have spotted molecules of glycolaldehyde – a simple form of sugar – around a young binary star with similar mass to the Sun, called IRAS 16923-2422. Glycolaldehyde ($C_2H_4O_2$) has been seen in interstellar space before, but this is the first time it has been found so near to a Sun-like star. The molecule is one of the ingredients in the formation of RNA, which – like DNA, to which it is similar – is one of the building blocks of life. The discovery shows that these building blocks are in the right place, at the right time, to be included in planets forming around the star.

For more information, see the press release: www.eso.org/public/news/eso1234

To learn more about ALMA, see:

Mignone C, Pierce-Price D (2010) The ALMA observatory: the sky is only one step away. *Science in School* 15: 44-49. www.scienceinschool.org/2010/issue15/alma

The European Southern Observatory (ESO) is by far the world's most productive ground-based astronomical observatory, with its headquarters in Garching near Munich, Germany, and its telescopes in Chile. ESO is the European partner in the ALMA project, which is a collaboration between Europe, North America and East Asia, in co-operation with the Republic of Chile. For more information, see: www.eso.org

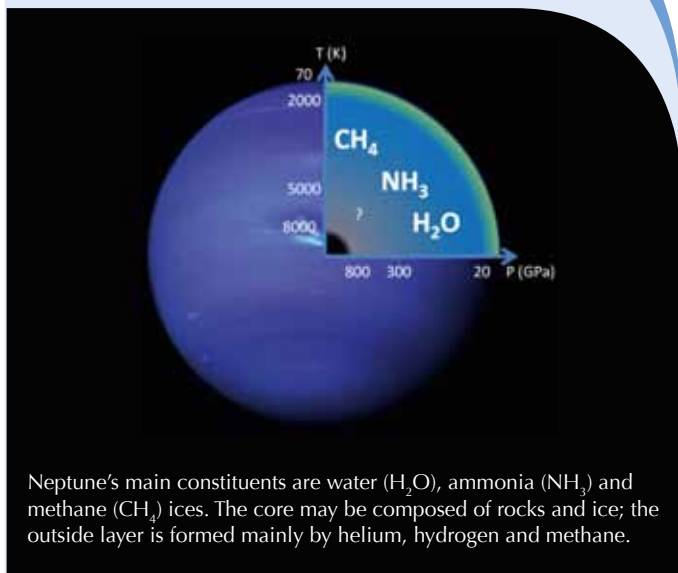
For a list of ESO-related articles in *Science in School*, see: www.scienceinschool.org/eso

The Rho Ophiuchi star-forming region seen in infrared light. IRAS 16293-2422 is the red object in the centre of the small square. Inset: an artist's impression of glycolaldehyde molecules, the form of sugar that has been found around IRAS 16293-2422.

Image courtesy of ALMA (ESO / NAOJ / NRAO) / L. Calçada (ESO) & NASA / JPL-Caltech / WISE Team



ESRF: Hot ice in Neptune



Neptune's main constituents are water (H_2O), ammonia (NH_3) and methane (CH_4) ices. The core may be composed of rocks and ice; the outside layer is formed mainly by helium, hydrogen and methane.

The main constituents of the planets Neptune and Uranus are water, ammonia and methane in solid phases called 'ices'. They are very different from ice on Earth, though: up to 5000 K hot and under extreme pressures of several million atmospheres. Under such conditions, scientists have predicted the existence of a new state of ammonia ice, called superionic ammonia. Superionicity is an exotic state of matter that behaves simultaneously as a crystal (fixed ion lattice) and as a liquid (diffusive ions).

Now scientists have confirmed the existence of superionic ammonia experimentally at the European Synchrotron Radiation Facility (ESRF): they submitted an ammonia sample to very high pressure while heating it, and followed its phase transformations using intense X-rays. Unexpectedly, however, they detected superionic ammonia at a lower temperature than predicted: 750 K instead of 1200 K. The exact boundaries of this superionic phase are crucial, because they determine whether it could exist in Neptune and Uranus. Therefore, the scientists will next test whether superionic ammonia is stable under even higher temperatures and pressures. Should this be the case, it could help explain the origin of the planets' magnetic fields, which are not yet well understood.

To learn more, see the news item on the ESRF website (www.esrf.eu), use the direct link (<http://tinyurl.com/superionic>), or read the research paper:

Ninet S, Datchi F, Saitta AM (2012) Proton disorder and superionicity in hot dense ammonia ice. *Physical Review Letters* 108(16): 165702-1–165702-5. doi: 10.1103/PhysRevLett.108.165702

Situated in Grenoble, France, ESRF operates the most powerful synchrotron radiation source in Europe. To learn more, see: www.esrf.eu

For a list of ESRF-related articles in *Science in School*, see: www.scienceinschool.org/esrf

Image courtesy of Sandra Ninet; image of Neptune by NASA / JPL-Caltech; model adapted from Hubbard WB, Podolak M, Stevenson DJ (1995) The interior of Neptune. In Cruikshank DP (ed) Neptune and Triton. pp. 109-138. Tucson, AZ, USA: University of Arizona Press. ISBN 978-0816515257

Image courtesy of European XFEL



Tuning the pole heights of an undulator in the magnetic measurement laboratory.

European
XFEL

European XFEL: The brightest light source on Earth

The European XFEL will deliver up to 27 000 very intense X-ray light flashes per second with a brightness more than 100 septillion (100×10^{24}) times that of an ordinary 60 W light bulb, making the new research facility the brightest light source on Earth. Deliveries of the devices generating this incredible firework – the undulators – to the site in Hamburg have been taking place since October 2012. The magnetic structures of the undulators will force accelerated electrons onto a slalom course, inducing them to emit X-ray flashes of extraordinary quality.

The new facility will have three undulator systems, with the two larger ones each 212 m long. They consist of segments that have been produced in close collaboration with European XFEL. Before the segments are installed, the undulator group of European XFEL will extensively measure and tune their magnetic properties. The light they produce will eventually be used, for example, to examine biomolecules, to film ultrafast processes and to study matter under extreme conditions.

The European X-ray Free Electron Layer (XFEL) is a research facility currently under construction in the Hamburg area in Germany. Its extremely intense X-ray flashes will be used by researchers from all over the world. To learn more, see: www.xfel.eu

For a list of *Science in School* articles relating to European XFEL, see: www.scienceinschool.org/xfel

To learn how to use this code, see page 65.



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To browse the other EIRO news articles, see: www.scienceinschool.org/eironews

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Image courtesy of schaubli / photoopia.com



Image courtesy of Dieter / pixelio.de



Image courtesy of John Carmemola / iStockphoto.com

ILL
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ILL: Cold-blooded platypus and hot-blooded chicken

Using the facilities at the Institut Laue-Langevin (ILL), a team of biologists have shown that haemoglobin in different species has evolved to perform its function as an oxygen carrier very effectively at that species' body temperature. They found variations in the amino-acid composition of this iron-rich protein, which is found in all vertebrates, that affect the protein's softness and flexibility, and thus its ability to withstand warmer or cooler temperatures.

The team studied haemoglobin from a range of vertebrates. The duck-billed platypus, at 33 °C, has the lowest body temperature of all endotherms (vertebrates that maintain a constant body temperature, such as mammals or birds). Humans, at 36.6 °C, have an intermediate body temperature, whereas the chicken, at 41 °C, is as hot-blooded as birds tend to be. The scientists also studied the ectothermic saltwater crocodile – which has a body temperature that is regulated by the environment, varying between 25 and 34 °C.

The scientists found a direct correlation between the resilience of haemoglobin and the average body temperature of the species from which it was sampled. In other words, each species' haemoglobin appears to have evolved to unfold at exactly the right body temperature.

To learn more, see the press release on the ILL website (www.ill.eu), use the direct link (<http://tinyurl.com/hbevolution>), or read the research paper:

Stadler AM et al. (2012) Thermal fluctuations of haemoglobin from different species: adaptation to temperature via conformational dynamics. *Journal of the Royal Society* 9(76): 2845-2855. doi: 10.1098/rsif.2012.0364

ILL is an international research centre at the leading edge of neutron science and technology, based in Grenoble, France. To learn more, see: www.ill.eu

For a list of ILL-related articles in *Science in School*, see: www.scienceinschool.org/ill

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